## C.U.SHAH UNIVERSITY Summer Examination-2016

## Subject Name : Heat and Thermodynamics

	Subject Code	: 4SC03PHC1	Branch : B.Sc. (All)			
	Semester : 3 Instructions:	Date : 30/04/2016	Time : 02:30 To 05:30	Marks : 70		
	<ul><li>(1) Use of</li><li>(2) Instruct</li><li>(3) Drawn</li><li>(4) Assum</li></ul>	Programmable calculate tions written on main an neat diagrams and figure he suitable data if needed	or & any other electronic instrument aswer book are strictly to be obeyed. s (if necessary) at right places.	is prohibited.		
0-1	Atte	mpt the following ques	tions:	(14)		
C	a) Writ	e statement of Zeroth lav	w of thermodynamics.	01		
	<b>b</b> ) Writ	e statement of Third law	of thermodynamics.	01		
	c) Defin	ne: Compressibility.		01		
	d) Wha	) What is expansion coefficient?				
	e) Defin	Define: Atmosphere Lapse rate.				
	f) Writ	e First law of thermodyn	amics.	01		
	g) Writ	e Clausius's statement o	f Second law of thermodynamics.	01		
	<b>h</b> ) Wha	t is radiant heat?		01		
	i) Wha	t is absorbing power?		01		
	j) Wha	t is reflecting power?		01		
	<b>k</b> ) Defin	ne: transmitting power.		01		
	I) Writ	e expression of Clausius	-Clapeyron relation.	01		
	m) Defin	ne: black body material.		01		
• • •	n) Writ	e statement of Wien's la	W.	01		
Atte	mpt any four q	uestions from Q-2 to Q	-8			
Q-2	Atte	mpt all questions	1 1 4 44	(14)		
	a) Expl	ain in detail $1^{-1}$ and $2^{-1}$ o	order phase transition	07		
	<b>D</b> ) Eluc	idate Carnot engine and	its efficiency.	07		
Q-3	Atte	mpt all questions		(14)		
	a) Defin	ne: specific heat. Develo	p the relation Cp-Cv=R.	07		
	b) Dete	rmine the formula of wo	rk done during an adiabatic process.	07		
Q-4	Atte	mpt all questions		(14)		
-	a) Disc	uss reversible and irreve	rsible changes in thermodynamic pro	ocess. 05		
	b) Expl	ain temperature-entropy	diagram.	05		
	c) Expl	ain heat and work as a st	tate function.	04		

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Q-5	Attempt all questions		
	a)	Explain thermal equilibrium with suitable example.	05
	b)	Explain Joule-Kelvin coefficient for ideal and Van der wall gases.	05
	c)	Write a short note on Stefan's law.	04
Q-6		Attempt all questions	
-	a)	Define: TdS equations. Develop $1^{st}$ and $2^{nd}$ TdS equations in thermodynamics.	07
	b)	Calculate the formula of the work done during isothermal process.	07
Q-7		Attempt all questions	(14)
-	a)	Describe derivations of Maxwell's relation.	07
	b)	Explain in detail main properties of radiant heat.	07
Q-8		Attempt all questions	(14)
-	a)	Originate $1^{st}$ and $2^{nd}$ order energy equations.	05
	b)	Derive differential form of First law of thermodynamics.	05
	c)	Discuss the energy distribution of black body.	04

